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# STRUCTURAL CHARACTERIZATIONS OF NBS<sub>2</sub> SINGLE CRYSTALS

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### ABSTRACT

Single crystals of niobium sulphide (NbS<sub>2</sub>) having a layered structure were grown by chemical vapour transport technique (CVT) using iodine as transporting agent. The stoichiometry of the grown crystals was confirmed on the basis of energy dispersive analysis by X-ray (EDAX) and the structural characterization was accomplished by X- ray diffraction (XRD) studies. The crystals were found to possess hexagonal crystal structure. The lattice parameters, volume, particle size and X-ray density was carried out for this crystal. The grown crystals were examined under optical zoom microscope for their surface topography study. Hall Effect measurements were carried out on grown crystals at room temperature. The positive value of Hall coefficient implies that this crystal is p-type in nature. The results obtained are discussed in detail.

Key Words: Crystal growth, semiconductor, EDAX, XRD

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### **INTRODUCTION**

Transition metal dichalcogenides having general formula TX<sub>2</sub>, where T is a transition metal and X is either S or Se, have been of considerable interest for the past two decades. They have unusual physical properties and can be intercalated with both Lewis bases and metals. The transition metal dichalcogenides, NbS<sub>2</sub> and NbSe<sub>2</sub> are layered compounds consisting of sandwiches with strong covalent/ionic intralayer bonds and weak van der Waals interlayer interactions. The stacking sequence of S and Nb are AcA BcB (2H- NbS<sub>2</sub> structure) or AbA BcB CaC (3R- MoS<sub>2</sub> structure) for NbSe<sub>2</sub> several more stacking sequences exist Lieth and Terhaell (1977); Boswell et al., (1976) and Harper (1977). Structure elucidation of the hexagonal platelets yielded crystal structures in the space group  $P6_3/mmc$ with NbS<sub>2</sub> layers, separated by intercalated alkali metals. The alkali metal niobium disulphides exhibit many interesting properties such as ionic/electronic conductivity McEwen and Sienko (1982) and superconductivity Gareh et al., (1995); Dunn and Glaunsinger (1988). Niobium disulfide has been synthesized using stoichiometric NbS<sub>2</sub> and has been investigated by X-ray powder diffraction, vapor pressure measurements, thermogravimetric analysis, differential scanning calorimetry and SQUID magnetometry Suzuki and Harima (2005). These materials except  $2H-NbS_2$  undergo the charge density wave (CDW) transition and then the metallic properties remain in the CDW phase Subba and Shafer (1979). Niobium disulfide (2H-NbS<sub>2</sub>) is a material with a layered structure in which niobium atoms are linked covalently to six sulfur atoms in a trigonal prismatic coordination Xingcai et al., (2004).

### MATERIALS AND METHODS

Single crystals of NbS2 were grown by a chemical vapour transport technique using iodine as transporting agent. Pure elements (purity 99.99%) of niobium and sulphur (purity 99.9%) in stoichiometric proportions and narrow capillary with  $I_2$  (2 mg/cc) of ampoule volume were sealed in an evacuated ( $10^{-5}$  torr) quartz ampoule for compound preparation. The sealed ampoule was introduced into dual zone furnace at a constant reaction temperature to obtain the charge of NbS<sub>2</sub>. The charge so prepared was rigorously shaken to ensure proper mixing of the constituents and kept in the furnace again under

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appropriate condition to obtain single crystals of  $NbS_2$ . The charge composition and growth conditions are summarized in Table 1.

The chemical composition of the as grown crystals was confirmed with the help of energy dispersive analysis by X-ray (EDAX). The stoichiometric proportions of as grown crystals as well as the EDAX results are shown in Table 2.

Close examination of the surface composed of layers helps a great deal in understanding the mechanism by which a crystal grow. Therefore, it was thought worthwhile to make surface characterization of these grown crystals by optical microscopy. The surface microstructures of as grown crystals were examined by computer added optical zoom microscope (Make: Carl Zeiss, Model Axiotech 100HD). The surface microstructure on the face of as grown crystal is shown in Figure 1, which shows layered surface structures. For X-ray diffraction (XRD) work, several small crystals were finely grinded with the help of an agate mortar and filtered through 106-micron sieve to obtain grains of nearly equal size. The X-ray diffractograms were taken with Philips X-ray diffrractometer (model: PW1820) employing  $CuK_{\alpha}$ radiation.

The room temperature resistivity of all crystals was determined using van derr Pauw's technique. The Hall Effect measurements were carried out to determine Hall coefficient, mobility and carrier concentration.

## **RESULTS AND DISCUSSION**

The chemical vapor transport technique was used to grow single crystals of  $NbS_2$  because it yields large single crystals with relative ease. The EDAX studies confirmed that  $NbS_2$  single crystals are stoichiometrycally perfect. The presence of hexagonal spirals on the face of the grown crystals suggests that growth involves a screw dislocation mechanism. The X-ray diffractogram of as grown crystal is shown in Figure 2. The pattern consists of well-defined sharp diffraction peaks, indicating good crystallinity of the specimen.

The density ' $\rho$ ' of the grown crystals was calculated by the formula,

$$\rho = \frac{\sum A}{VN} \tag{1}$$

Where  $\sum$  A is the total weight of the atoms in the unit cell = MZ

M is the molecular weight and Z is the number of molecules/unit cell,

N is the Avogadro number and V is the volume of the unit cell for the hexagonal system given by

$$\frac{\sqrt{3}}{2}a^2c \,\mathring{A}^3 = 0.866 \,a^2c \,\mathring{A}^3 \qquad \dots \qquad (2)$$

The values of lattice parameters a (Å), b (Å), c (Å), X-ray density  $\rho$  and unit cell volume V (Å<sup>3</sup>) obtained from the diffraction data for as grown crystal of NbS<sub>2</sub> are presented in Table 3.

In order to obtain an idea about the grain size distribution in  $NbS_2$  single crystal, the particle size was calculated using Scherrer's formula given as

$$t = \frac{k \lambda}{\beta_{2\theta} \cos \theta_0} \tag{3}$$

Where t is the crystallite thickness as measured perpendicular to the reflecting plane; k is Scherrer's constant whose value is chosen as unity assuming the particles to be spherical;  $\lambda$  is the wavelength of the X-ray radiation;  $\beta_{2\theta}$  is the width at half the maximum intensity measured in radians, and  $\theta_0$  is the Bragg angle. A Table 4 records the crystallite size of grown crystal.

In the case of hexagonal close packed metals, it is possible to make a realistic estimation of the growth fault probability ' $\alpha$ ' and the deformation fault probability ' $\beta$ ' by measuring the half width of X-ray diffraction lines. Reflections for which h - k = 3n where 'n' is an integer, are independent of stacking

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faults whereas reflections for which  $h - k = 3n \pm 1$  and  $l \neq 0$  depend upon the faults in the crystal structure. An estimation of the deformation and growth fault probability can be obtained from the following formula for (h k l) values with 'l' even

$$(3\alpha + 3\beta) = \frac{\beta_{2\theta} \times \pi^2 \times c^2}{360 \times l \times d^2 \times \tan \theta}$$
(4)

where  $\beta_{2\theta}$  is the full width at half the maximum intensity expressed in degrees,  $c = d_{002}$ , l is the Miller index in the (h k l) plane for which the estimation of ' $\alpha$ ' and ' $\beta$ ' is being made, 'd' is the inter planer spacing for (h k l) reflection in question,  $\theta$  is the Bragg angle corresponding to this (h k l) plane. The formula for (h k l) values with 'l' odd is given as

$$(3\alpha + \beta) = \frac{\beta_{2\theta} \times \pi^2 \times c^2}{360 \times l \times d^2 \times \tan \theta}$$
(5)

From the equations 4 and 5, it is clear that by measuring the half width  $\beta_{\theta}$  for reflections with both even and odd values of 'l' it is possible to calculate the stacking fault probabilities  $\alpha \& \beta$ . In calculating the half width of the reflections, instrumental broadening is neglected. The results of estimation of  $\alpha \& \beta$  are given in Table 5. The Hall coefficient, carrier concentration and mobility of as grown crystals are listed in Table 6. The positive sign of Hall coefficients suggests that all crystals are p-type nature.

Table 1: Growth parameters of NbS2 single crystal	l grown	using chemical	vapors	transport
technique				

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	D (1	Growth <sup>–</sup> Temperature	Physical Characteristics of the Crystals			
Sample	Reaction Temperature		Growth Time hr	Plate Area mm <sup>2</sup>	Thickness mm	Color and Appearance
NbS <sub>2</sub>	1083	850	240	6	0.03	Black Shining
Table 2: The EDAX data NbS2 single crystals						
Sam	S ple	Stoichiometric Proportion (Wt %)		EDAX Results (Wt %)		
	- <u> </u>	Nb		Nb		S
Nbs	S <sub>2</sub> 59	.16	40.84	58.11 41.89		41.89
Table 3: Structural parameters for NbS2 single crystals						
Sample	$\mathbf{a} = \mathbf{b} \left( \mathbf{A} \right)$	a = b (Å) Reported	c (Å)	c (Å) Reported	Volume (Å <sup>3</sup> )	X-Ray Density (gm/cm <sup>3</sup> )
NbS <sub>2</sub>	3.33	3.33[11]	17.86	17.86 <sup>10</sup>	171.8	3.03

Table 4: The h k l reflections, d- spacing, 2 $\theta$  Values, Peak Intensity,  $\beta_{2\theta}$  value and particle size for NbS<sub>2</sub> single crystals

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(h k l)	d-Spacing	Angle 2θ (Degree)	Peak Intensity (cont/sec)	Tip Width β <sub>2θ</sub>	Particle Size (Å)
101	2.83	31.58	109.27	0.42	425.53
102	2.73	32.73	101.29	0.42	426.77
103	2.58	34.62	8.11	0.60	462.52
204	1.36	68.43	23.95	0.42	762.55

Table 5: Estimation of stacking fault probability of NbS <sub>2</sub> single crystals				
(h k l)	$3\alpha + 3\beta$	$3\alpha + \beta$	α	β
101	_	0.007		
102	0.838	-	0.022	0.400
103	-	0.847	0.033	0.499
204	0.729	-		

Table 6: The Hall coefficient, mobility and carrier concentration of NbS <sub>2</sub> single crystal at room
Temperature

Sample	Resistivity r (W·cm)	Conductivity s (W·cm) <sup>-1</sup>	Hall Coefficient $R_H$ (cm <sup>3</sup> /coul.)	Mobility µ (cm²/Vs)	Carrier Concentration N <sub>c</sub> (cm) <sup>-3</sup>
NbS <sub>2</sub>	1.18	0.84	553.8	469.3	$1.13 \times 10^{16}$



## CONCLUSION

- It is possible to grow large size single crystals of  $NbS_2$  by chemical vapour transport technique using iodine as a transporting agent. A single crystal of NbS2 was found to grow in form of thin platelets showing a mirror like metallic luster.
- Growth and deformation fault probability are found which shows the layer structure of as grown crystals.
- From EDAX analysis stoichiometry of as grown crystal of NbS<sub>2</sub> is nearly preserved.
- X-ray diffraction analysis shows that the grown crystal possesses hexagonal structure.
- Surface morphology reveals the layer structure of these crystals.

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