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THE EDGE-SZEGED INDEX OF THE POLYCYCLIC AROMATIC HYDROCARBONS PAH_K

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ABSTRACT

Let G=(V,E), be a simple connected molecular graph with the vertex set V=V(G) and the edge set E=E(G). The PI index of G is defined as $PI_e(G)=\sum_{e=uv\in E(G)} [m_u(e/G)+m_v(e/G)]$. The number of edges

of G lying closer to u than to v is denoted by $m_u(e/G)$ and the number of edges of G lying closer to v than to u is denoted by $m_v(e/G)$. In this paper, we compute a closed formula of the PI index of molecular graph "Polycyclic Aromatic Hydrocarbons PAH_k ", for all positive integer number k.

Keywords: Molecular Graph, Polycyclic Aromatic Hydrocarbons PAHk; PI index PIe.

INTRODUCTION

Let G=(V,E), be a simple connected molecular graph with the vertex set V=V(G) and the edge set E=E(G). In molecular graph G, vertices are corresponding to the atoms and edges corresponding to the bonds. An edge e=uv ($\in E(G)$) of graph G is joined between two vertices u and v ($\in V(G)$). A general reference for the notation in graph theory is (West, 1996; Todeschini and Consonni, 2000; Trinajstić, 1992; Wiener, 1947).

A topological index is a real number related to a molecular graph. It must be a structural invariant, i.e., it does not depend on the labeling or the pictorial representation of a graph. The Wiener index W is the first topological index proposed to be used in chemistry. It was introduced in 1947 by Wiener for characterization of alkanes. The Wiener is defined as the sum of all distances between distinct vertices as:

$$W(G) = \frac{1}{2} \sum_{u \in V(G)} \sum_{v \in V(G)} d(u, v)$$

where d(u,v) be the distance (the number of edges in a shortest path) between vertices u and v of G.

The PI index PI_e introduced by Khadikar (2000, 2001, and 2002). It is defined as the sum of $m_u(e/G)+m_v(e/G)$ between all edges e=uv of a graph G, where $m_u(e/G)$ is the number of edges of G lying closer to u than to v and $m_v(e/G)$ is the number of edges of G lying closer to v than to u. In other words, the PI index $PI_e(G)$ is equal to:

$$PI_{e}(G) = \sum_{e=uv \in E(G)} [m_{u}(e/G) + m_{v}(e/G)].$$

Mathematical properties of the PI index for some classes of chemical graphs can be found in recent papers (Ashrafi and Loghman, 2006; Ashrafi and Vakili-Nezhad, 2006; Ashrafi and Loghman, 2006; Ashrafi and Loghman, 2006; Ashrafi and Rezaei, 2007; Deng, 2006; Farahani, 2013; Farahani, 2013; Manoochehrian *et al.*, 2007).

In this paper the PI index PI_e of one of famous hydrocarbon molecules "the *Polycyclic Aromatic Hydrocarbons PAH*_k" are computed.

RESULTS AND DISCUSSIONS

Main Results and Discussions

Let PAH_k be the "Polycyclic Aromatic Hydrocarbons" for all positive integer number k and its general representation is shown in Figure 1. The hydrocarbons molecule "Polycyclic Aromatic Hydrocarbons" is more practical in the chemical and physics and play a role in graphitization of organic materials. In Figure

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1, the general case of this hydrocarbons molecule are shown. For further study of its properties, see paper series (Wiersum and Jenneskens, 1997; Berresheim *et al.*, 1999; Bauschlicher and Bakes, 2000; Brand *et al.*, 1998; Morgenroth *et al.*, 1998; Dtz *et al.*, 2000; Yoshimura *et al.*, 2001; Stein and Brown, 1987; Dietz *et al.*, 2000; Huber *et al.*, 2003; Jug and Bredow, 2004; Farahani, 2013; Farahani, 2013; Farahani, 2015; Farahani, 2015; Farahani and Schultz, 2013; Farahani and Gao, 2015; Gao and Farahani, 2015). The PI index $PI_e(PAH_k)$ of *Polycyclic Aromatic Hydrocarbons PAH_k* is presented in the following theorem.

Theorem 1: The PI index PI_e of Polycyclic Aromatic Hydrocarbons $PAH_k(\forall k \geq 1)$ is equal to $PI_e(PAH_k) = 81k^4 + 40k^3 + 6k^2 - k$.

Proof: Let PAH_k be the Polycyclic Aromatic Hydrocarbons ($\forall k \geq 1$). From the general representation of Polycyclic Aromatic Hydrocarbons PAH_k in Figure 1, one can see that there are $6k^2$ Carbon atoms and 6k Hydrogen atoms in the structure of PAH_k . Therefore, the number of vertices/atoms of PAH_k is equal to $6k^2+6k$ and alternatively PAH_k have $\frac{1}{2}(1\times 6k+3\times 6k^2)=9k^2+3k$ edge/bonds.

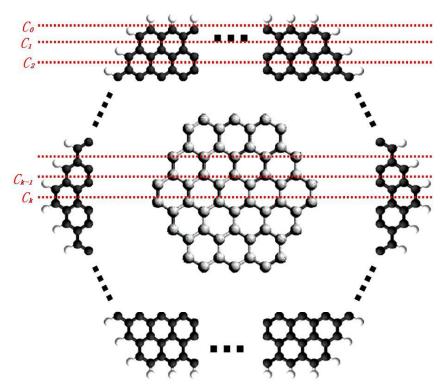


Figure 1: All orthogonal cuts of the Polycyclic Aromatic Hydrocarbons PAH_k .

By definition of the PI index PI_e , it is enough to compute $m_u(e/PAH_k)$ and $m_v(e/PAH_k)$. So, to compute $m_u(e/PAH_k)$ and $m_v(e/PAH_k)$ for all edge $e=uv \in E(PAH_k)$, we attend to Figure 1. By Figure 1, we see that an arbitrary edge cut $C_i(i=0,...,k)$, cut k+i edge of PAH_k and also, for ith cut C_i ;

$$m_{\nu}(e|PAH_k) = 2(k+i) + (k+(i-1)) + 2(k+(i-1)) + (k+(i-2)) + \dots + 2(k+1) + k = \sum_{s=i}^{1} 2(k+(i+1)-s)$$

$$+ \sum\nolimits_{s=1}^{i} (k+(i-s)) = \sum\nolimits_{s=1}^{i} \left(\left(3k + 3i + 2 \right) - 3s \right) = (3k+3i+2)i - \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \left(3k + \frac{1}{2} \right)i \cdot \frac{3}{2}i(i+1) = \frac{3}{2}i^{2} + \frac{3}{2}i^{2} +$$

Also, it is easy to see that $\forall i=0,...,k; |E(PAH_k)|=m_v(e/PAH_k)+m_u(e/PAH_k)+|C_i|$ and $|C_i|=k+i$, thus $m_u(e/PAH_k)=9k^2+3k-(k+i)-(3/2i^2+(3k+1/2)i^2)=9k^2+2k-3/2i^2-(3k+3/2)i$.

Now by according to the definition of PI index, we have the following computations of the Polycyclic Aromatic Hydrocarbons $PI_e(PAH_k)$.

$$PI_e(PAH_k) = \sum_{e=uv \in E(PAH_k)} m_v(e/PAH_k) + m_u(e/PAH_k)$$

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$$= \sum_{\substack{e \Rightarrow vu \in C(e_i) \\ i=1,\dots,k-1}} 6(k+i)(m_v(e|PAH_k) + m_u(e|PAH_k)) + \sum_{e_k \in C(e_k)} 3(2k)(m_v(e|PAH_k) + m_u(e|PAH_k))$$

$$= 6(k)(0+9k^2+2k) + 6(k+1)(3k+2+9k^2-k-3) + \dots + 6(2k-1)(9/2k^2-7/2k+1+9/2k^2+9/2k)$$

$$+ 6(2k)(9/2k^2+1/2k)$$

$$= \sum_{i=0}^k 6(k+i)(9k^2+2k-i) - (6k)(9k^2+k))$$

$$= 6\sum_{i=0}^k (9k^3+2k^2) + 6\sum_{i=0}^k i(9k^2+k) - 6\sum_{i=0}^k i^2-6k^2(9k+1)$$

$$= 6k^2(k+1)(9k+2) + 3k^2(k+1)(9k+1) - 6(\frac{k(k+1)(2k+1)}{6}) - 6k^2(9k+1)$$

$$= 6k^2(9k^2+11k+2) + 3k^2(9k^2+10k+1) - (2k^3+3k^2+k) - 6k^2(9k+1)$$

$$= 81k^4 + 40k^3 + 6k^2 - k.$$

Therefore, the proof of Theorem 1 is completed.■

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REFERENCES

Ashrafi AR and Loghman A (2006). Padmakar-Ivan Index of TUC4C8(S) Carbon Nanotubes. *Journal of Computational and Theoretical Nanoscience* **3** 378-381.

Ashrafi AR and Loghman A (2006). PI Index of Armchair Polyhex Nanotubes. *Ars Combinatoria* **80** 193-199.

Ashrafi AR and Loghman A (2006). PI Index of Zig-Zag Polyhex Nanotubes. *MATCH:* Communications in Mathematical and in Computer Chemistry **55** 447–452.

Ashrafi AR and Rezaei F (2007). PI Index of Polyhex Nanotori. *MATCH: Communications in Mathematical and in Computer Chemistry* 57 243-250.

Ashrafi AR and Vakili-Nezhad GR (2006). Computing the PI Index of some Chemical Graph Related to Nanostructures. *Journal of Physics: Conference Series* **29** 181–184.

Bauschlicher CW and Jr Bakes ELO (2000). Infrared spectra of polycyclic aromatic hydrocarbons (PAHs). *Chemical Physics* **262** 285-291.

Berresheim J, Müller M and Müllen K (1999). Polyphenylene nanostructures. *Chemical Reviews* **99**(7) 1747–1786.

Brand JD, Geerts Y, Mullen K, Craats AM and Warman JM (1998). Rapid charge transport along self-assembling graphitic nano-wires. *Advanced Materials* **10** 36-38.

Deng H (2006). Extremal Catacondensed Hexagonal Systems with Respect to the PI Index. *MATCH: Communications in Mathematical and in Computer Chemistry* **55** 453–460.

Dietz F, Tyutyulkov N, Madjarova G and Müllen K (2000). Is 2-D graphite an ultimate large hydrocarbon? II. structure and energy spectra of polycyclic aromatic hydrocarbons with defects. *The Journal of Physical Chemistry B* **104**(8) 1746-1761.

Dtz F, Brand JD, Ito S, Ghergel L and Müllen K (2000). Synthesis of large polycyclic aromatic hydrocarbons: variation of size and periphery. *Journal of the American Chemical Society* **122**(32) 7707-7717.

Farahani MR (2013). Computing Edge-PI index and Vertex-PI index of Circumcoronene series of Benzenoid H_k by use of Cut Method. *International Journal of Mathematical Modeling and Applied Computing* 1(6) 41-50.

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Farahani MR (2013). Hosoya, Schultz, Modified Schultz Polynomials and Their Topological Indices of Benzene Molecules: First Members of Polycyclic Aromatic Hydrocarbons *PAH_k*. *International Journal of Theoretical Chemistry* **1**(2) 09-16.

Farahani MR (2013). Some connectivity indices of polycyclic aromatic hydrocarbons PAH_k . Advances in Materials and Corrosion 1 65-69.

Farahani MR (2013). Using the Cut Method to Computing Edge Version of Co-PI Index of Circumcoronene Series of Benzenoid H_k . Pacific Journal of Applied Mathematics 5(3) 65-72.

Farahani MR (2013). Vertex Version of Co-PI Index of Circumcoronene Series of Benzenoid H_k . *Polymers Research Journal* 7(3).

Farahani MR (2013). Zagreb indices and Zagreb polynomials of polycyclic aromatic hydrocarbons. *Journal of Chemica Acta* **2** 70-72.

Farahani MR (2014). Schultz and Modified Schultz Polynomials of Coronene Polycyclic Aromatic Hydrocarbons. *International Letters of Chemistry, Physics and Astronomy* **13**(1) 1-10.

Farahani MR (2015). Exact Formulas for the First Zagreb Eccentricity Index of Polycyclic Aromatic Hydrocarbons *PAH_k*. *Journal of Applied Physical Science International* **4**(3) 185-190.

Farahani MR (2015). The Second Zagreb Eccentricity Index of Polycyclic Aromatic Hydrocarbons *PAH_k*. *Journal of Computational Methods in Molecular Design* **5**(2) 115-120.

Farahani MR and Gao W (2015). On Multiple Zagreb indices of Polycyclic Aromatic Hydrocarbons PAH. *Journal of Chemical and Pharmaceutical Research* **7**(10) 535-539.

Gao W and Farahani MR (2015). Theta polynomial $\Theta(G,x)$ and Theta index $\Theta(G)$ of Polycyclic Aromatic Hydrocarbons PAH_k . *Journal of Advances in Chemistry*, In press.

Huber SE, Mauracher A and Probst M (2003). Permeation of low-Z atoms through carbon sheets: density functional theory study on energy barriers and deformation effects. *Chemistry - A European Journal* 9 2974-2981.

Jug K and Bredow T (2004). Models for the treatment of crystalline solids and surfaces. *Journal of Computational Chemistry* **25** 1551-1567.

Khadikar PV (2000). On a Novel Structural Descriptor PI. National Academy Science Letters 23

Khadikar PV, Kale PP, Deshpande NV, Karmarkar S and Agrawal VK (2001). Novel PI Indices of Hexagonal Chains. *Journal of Mathematical Chemistry* **29** 143–150.

Khadikar PV, Karmarkar S and Varma RG (2002). The Estimation of PI Index of Polyacenes. *Acta Chimica Slovenica* **49** 755–771.

Manoochehrian B, Yousefi-Azari H and Ashrafi AR (2007). PI Polynomial of some Benzenoid Graphs. MATCH: Communications in Mathematical and in Computer Chemistry 57 653-664

Morgenroth F, Kübel C, Müller M, Wiesler UM, Berresheim AJ, Wagner M and Müllen K (1998). From three-dimensional polyphenylene dendrimers to large graphite subunits. *Carbon* 36(5-6) 833-837.

Stein SE and Brown RL (1987). Pi.-Electron properties of large condensed polyaromatic hydrocarbons. *Journal of the American Chemical Society* 109(12) 3721-3729.

Todeschini R and Consonni V (2000). Handbook of Molecular Descriptors (Wiley, Weinheim).

Trinajstić N (1992). Chemical Graph Theory (CRC Press, Boca Raton, FL).

West DB (1996). An Introduction to Graph Theory (Prentice-Hall).

Wiener H (1947). Structural determination of the paraffin boiling points. *Journal of American Chemical Society* **69** 17–20.

Wiersum UE and Jenneskens LW (1997). *Gas Phase Reactions in Organic Synthesis*, edited by Valle Y (Gordon and Breach Science Publishers, Amsterdam, The Netherlands) 143–194.

Yoshimura K, Przybilla L, Ito S, Brand JD, Wehmeir M, Rder HJ and Müllen K (2001). Characterization of Large Synthetic Polycyclic Aromatic Hydrocarbons by MALDI- and LD-TOF Mass Spectrometry. *Macromolecular Chemistry and Physics* 202 215-222.